

# A Theoretical Study of Pyrolysis of JP-10 (*exo*-Tetrahydrodicyclopentadiene) and its Primary and Secondary Unimolecular Decomposition Products

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**Abstract.** Theoretical calculations of the rate constants and product branching ratios in the pyrolysis of *exo*-tetrahydrodicyclopentadiene (JP-10) and its initial decomposition products at combustion-relevant pressures and temperatures have been performed and compared to experimental results from recently reported molecular beam photoionization mass spectrometry study of the pyrolysis of JP-10 (Zhao, L. et al. *Phys. Chem. Chem. Phys.* **2017**, *19*, 15780). The results allow us to quantitatively assess the decomposition mechanisms of JP-10 by a direct comparison with the nascent product distribution - including radicals and thermally labile closed-shell species - detected in the short-residence-time molecular beam photoionization mass spectrometry experiment. In accord with the experimental data, the major products predicted by the theoretical modeling include methyl radical (CH<sub>3</sub>), acetylene (C<sub>2</sub>H<sub>2</sub>), vinyl radical (C<sub>2</sub>H<sub>3</sub>), ethyl radical (C<sub>2</sub>H<sub>5</sub>), ethylene (C<sub>2</sub>H<sub>4</sub>), allyl radical (C<sub>3</sub>H<sub>5</sub>), 1,3-butadiene (C<sub>4</sub>H<sub>6</sub>), cyclopentadienyl radical (C<sub>5</sub>H<sub>5</sub>), cyclopentadiene (C<sub>5</sub>H<sub>6</sub>), cyclopentenyl radical (C<sub>5</sub>H<sub>7</sub>), cyclopentene (C<sub>5</sub>H<sub>8</sub>), fulvene (C<sub>6</sub>H<sub>6</sub>), benzene (C<sub>6</sub>H<sub>6</sub>), toluene (C<sub>7</sub>H<sub>8</sub>), and 5-methylene-1,3-cyclohexadiene (C<sub>7</sub>H<sub>8</sub>). We found that ethylene, allyl radical, cyclopentadiene, and cyclopentenyl radical are significant products at all combustion-relevant conditions, whereas the relative yields of the other products depend on temperature. The most significant temperature trends predicted are increasing yields of the C<sub>2</sub> and C<sub>4</sub> species and decreasing yields of the C<sub>1</sub>, C<sub>6</sub>, and C<sub>7</sub> groups with increasing temperature. The calculated pressure effect on the rate constant for the decomposition of JP-10 via initial C-H bond cleavages becomes significant at temperatures above 1,500 K. The initially produced radicals will react away to form closed-shell molecules, such as ethylene, propene, cyclopentadiene, cyclopentene, fulvene, and benzene, which were observed as the predominant fragments in the long-residence-time experiment. The calculated rate constants and product branching ratios should prove useful to improve the existing kinetic models for the JP-10 pyrolysis.